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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.059 wR factor = 0.167 Data-to-parameter ratio = 15.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[2-(Benzotriazol-1-yl)ethyl][2-(benzotriazol-2-yl)ethyl]amine

In the title compound, $C_{16}H_{17}N_7$, the dihedral angle between the two benzotriazole ring systems is 73.97 (4)°. N-H···N and C-H···N hydrogen bonds link the molecules into a chain along the *a* axis. The packing is further stabilized by $\pi-\pi$ stacking interactions involving two benzotriazole ring systems. Received 30 August 2005 Accepted 19 September 2005 Online 24 September 2005

Comment

Benzotriazole and its derivatives have good anti-wear properties, and they are well established corrosion inhibitors for copper and its alloys (Ren *et al.*, 1994; Dugdale & Cotton, 1963). In view of these properties, the title compound, (I), was synthesized according to the literature procedure (Liao *et al.*, 1998), and its crystal stucture is reported here.



The structural investigation of (I) indicates that the geometry around atom N4 is pyramidal (Fig. 1). The Csp^3-N bond distances range from 1.455 (3) to 1.460 (2) Å (Table 1). The dihedral angle between the two benzotriazole ring systems is 73.97 (4)°. The N $-Csp^3-Csp^3-N$ torsion angles [179.4 (2) and 58.6 (2)°] define the conformation of the central linkage.

In the crystal packing of (I), molecules translated one unit cell along the *a* direction are linked to form a chain *via* N— H···N and C—H···N hydrogen bonds (Table 2 and Fig. 2). The packing is further stabilized by π - π stacking interactions between the N1-benzotriazole ring system at (x, y, z) and the N5-benzotriazole ring system at the symmetry position $(x - 1, \frac{1}{2} - y, z - \frac{1}{2})$ [centroid···centroid distance 3.718 (2) Å].



© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved The structure of (I). Displacement ellipsoids are drawn at the 30%

Experimental

Compound (I) was synthesized according to the literature procedure of Liao *et al.* (1998). Single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution at room temperature.

 $D_r = 1.339 \text{ Mg m}^{-3}$

Cell parameters from 5137

3313 independent reflections

2647 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\theta = 2.2-26.6^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$

T = 293 (2) K

 $R_{\rm int}=0.029$

 $\theta_{\rm max} = 27.0^{\circ}$

 $h = -7 \rightarrow 7$ $k = -31 \rightarrow 31$

 $l = -12 \rightarrow 12$

Block, colourless $0.40 \times 0.30 \times 0.20$ mm

Crystal data

 $\begin{array}{l} C_{16}H_{17}N_7 \\ M_r = 307.37 \\ \text{Monoclinic, } P2_1/c \\ a = 6.2180 \ (16) \text{ Å} \\ b = 24.893 \ (6) \text{ Å} \\ c = 10.020 \ (3) \text{ Å} \\ \beta = 100.650 \ (4)^{\circ} \\ V = 1524.2 \ (7) \text{ Å}^3 \\ Z = 4 \end{array}$

Data collection

Bruker SMART APEX CCD areadetector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 2003) $T_{\min} = 0.966$, $T_{\max} = 0.983$ 12667 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0783P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.059$	+ 0.5802P]
$wR(F^2) = 0.167$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.001$
3313 reflections	$\Delta \rho_{\rm max} = 0.71 \ {\rm e} \ {\rm \AA}^{-3}$
212 parameters	$\Delta \rho_{\rm min} = -0.34 \ {\rm e} \ {\rm \AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

C1-N1	1.363 (3)	C10-N6	1.459 (2)
C6-N3	1.369 (3)	C11-N5	1.352 (3)
C7-N1	1.460 (2) 1.470 (3)	C16-N7 N1-N2	1.353 (2) 1.345 (3)
C7-C8			
C8-N4	1.458 (3)	N2-N3	1.302 (2)
C9-N4	1.455 (3)	N5-N6	1.321 (2)
C9-C10	1.503 (3)	N6-N7	1.317 (2)
N1-C7-C8-N4	179.4 (2)	C10-C9-N4-C8	167.99 (17)
N4-C9-C10-N6	58.6 (2)	C7-C8-N4-C9	92.9 (3)

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
$\begin{array}{c} C8 - H8B \cdots N3^{i} \\ N4 - H4A \cdots N2^{i} \end{array}$	0.97	2.59	3.495 (3)	156
	0.92 (3)	2.60 (3)	3.254 (3)	128 (2)

Symmetry code: (i) x + 1, y, z.



Figure 2

Part of the crystal packing of (I), showing a hydrogen-bonded (dashed lines) chain along the *a* axis and π - π interactions. For clarity, only H atoms involved in the hydrogen bonding are shown. Atoms labelled with the suffixes a, b and c are generated by the symmetry operations (1 + x, y, z), (x - 1, y, z) and $(x - 1, \frac{1}{2} - y, z - \frac{1}{2})$, respectively.

Atom H4A was located in a difference Fourier map and refined isotropically. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with $C-H_{methylene} = 0.97$ Å and $C-H_{aromatic} = 0.93$ Å, and with $U_{iso}(H) = 1.2U_{eq}(C)$. The three highest peaks in the final difference map indicated disorder for the C7–C8–N4 chain segment. However, refinement based on a disordered chain segment model led to very high displacement parameters.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PLATON*.

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